

I'm not a robot!

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Flame Test Lab

*****The lab handout from class looks a little different than this handout, but the table and the questions are the same.

Observations & Data:

Table 1: Flame Test Observations

solution	metal ion present	colored streak of flame	written description of color(s) observed
NaCl			
KCl			
CaCl ₂			
SrCl ₂			
CuCl ₂			
unknown			

Calculations & Conclusions:

- Discuss some of the limitations of using flame tests to identify ions present in solution.
- Propose at least one possible method for improving the accuracy of results in this laboratory experiment.
- A green line of wavelength 546 nm is observed in the emission spectrum of hydrogen. Calculate the energy of one photon of this green light. Use the following mathematical relationships:

(Where c = frequency, λ = speed of light, ν = wavelength, h = Planck's constant, E = energy)

$$w = \frac{c}{\lambda} \quad E = h\nu$$

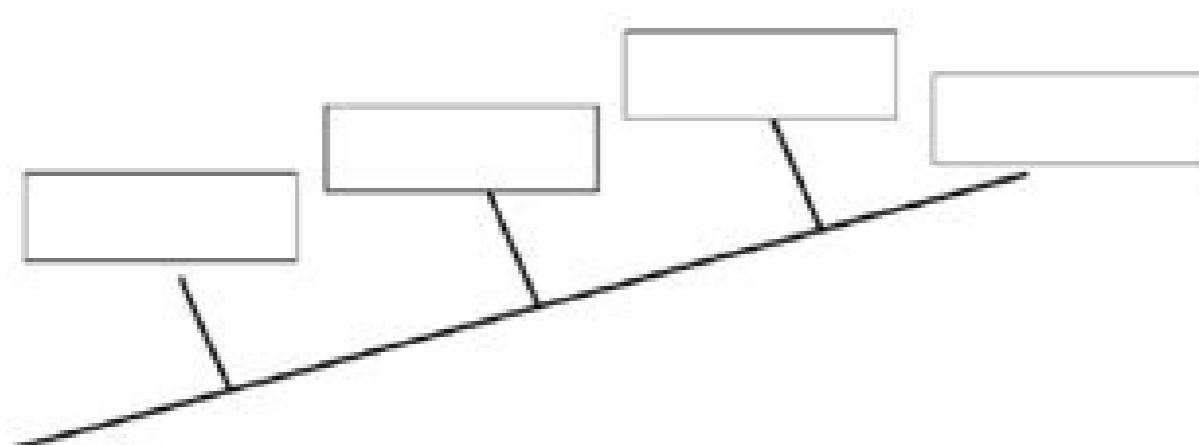
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Cladogram Worksheet

Fill in the following table. Mark an "X" if an organism has the trait.

	cells	legs	6 legs	wings
Worm				
Spider				
Carpenter Ant (black)				
Fly				

- Add each of these organisms to the cladogram below: worm, spider, ant, fly
- USING complete sentences, explain why you put each organism where you did on the cladogram.



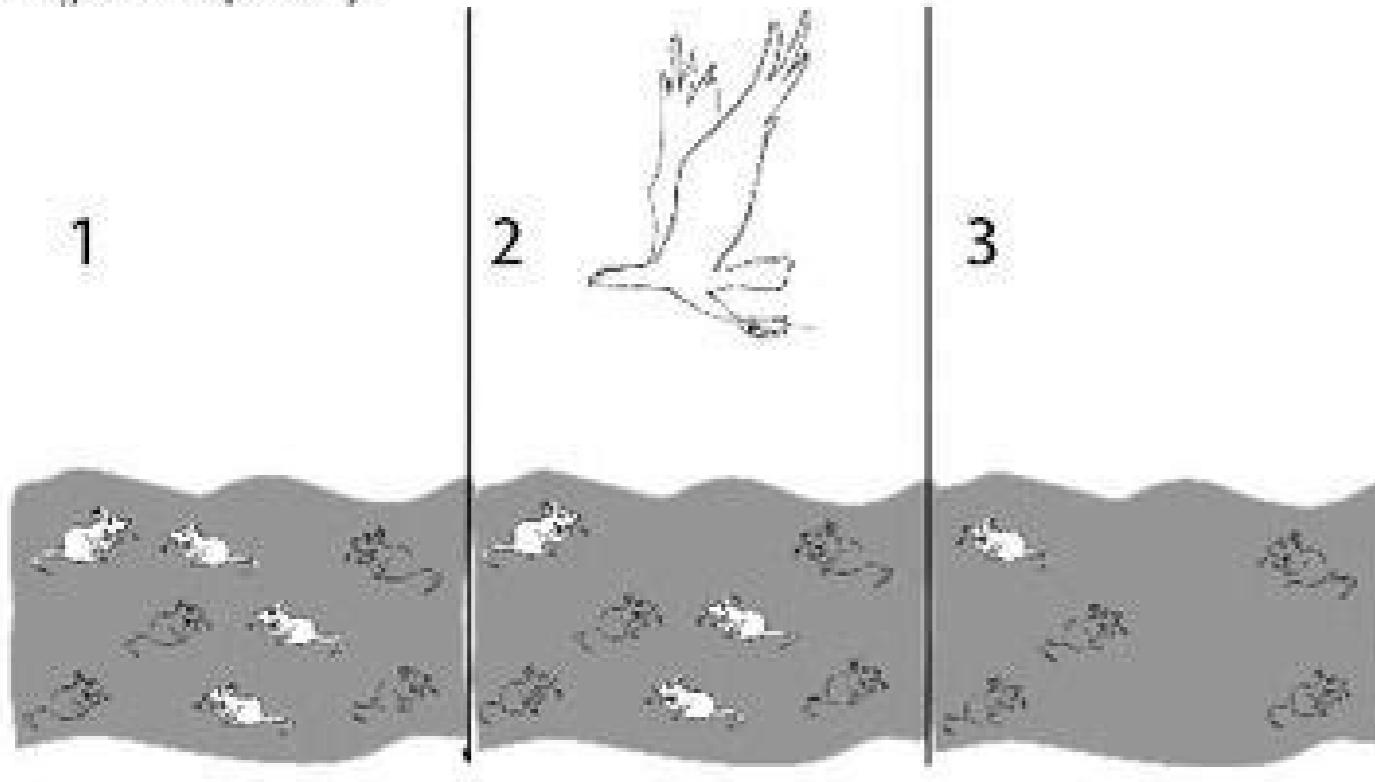
- On the cladogram above, add traits that make the organisms different from each other.
- According to your cladogram, which two species are more closely related: worms and spiders or worms and ants? How do you know?
- According to your cladogram, what species are flies most closely related to? How do you know?

Evolution by Natural Selection

Adapted from the University of California, Los Angeles Life Sciences 1 Demonstration Manual.
Copyright 2012 by Dr. Jennifer Doherty and Ngaiid Walker, Department of Biology, University of Pennsylvania.

For more information about the study, please contact Dr. Michael J. Koenig at (415) 502-2555 or via email at koenig@berkeley.edu.

★ Describe what is happening in figures 1-3. Is the population of mice different in figure 3 than in figure 1? Explain why.



Living things that are well adapted to their environment survive and reproduce. Those that are not well adapted don't survive and reproduce. An **adaptation** is any characteristic that increases fitness, which is defined as the ability to survive and reproduce.

★ What characteristic of the mice in figure 1 was an adaptation that increased fitness?

The table describes four female mice that live in a beach area which is mostly tan sand with scattered plants.

Characteristics of each female mouse	Color of Fur			
	Black	Tan	Tan and Black	Cream
Running speed	8 cm/sec.	6 cm/sec.	7 cm/sec.	5 cm/sec.
# pups produced by each female	0	11	3	0
Age at death	2 months	8 months	4 months	2 months

★ According to the definition given for fitness, which mouse would biologists consider the fittest? Explain why this mouse would be the fittest.

★ If a mouse's fur color is generally similar to its mother's color, what color fur would be the most common among the pups?

¹ Teachers are encouraged to copy this student handout for classroom use. A Word file (which can be used to prepare a modified version if desired), Teacher Preparation Notes, comments, and the complete list of our hands-on activities are available at <http://www.edu.brynmawr.edu/edc/eduactivities>.

Name: _____ M.I.: _____ Date: _____

Plane Test List

100

To observe the characteristic colors produced by certain metallic ions when vaporized in a flame; and to identify an unknown metal ion by means of the Bausch test.

Matthew B. Beckmann

- various origins
 - Human humor
 - 7 different methods in addition
 - spectroscopy
 - water broken with water

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- Procedure:**

 - Fill a beaker two-thirds of the way with water to dispose of used wooden splints.
 - Set up Bunsen burner. Adjust the gas and oxygen flow to produce a hot blue flame.
 - Take a wooden splint from the metal ion solution and insert into flame. Before the wooden splint begins to burn, record the color of the flame and the using the spectrometer record the wavelengths of the metal's visible atomic spectrum.
 - Place the wooden splint in the ashtray holder for at least one minute before disposal of splint or trash can.
 - Repeat steps 1 through 4 for all 6 ions.
 - Obtain a splint from the unknown solution. Repeat steps 1 through 4.

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Table 1 – Observed Flame Colors & Visible Atomic Spectrum Wavelengths

Solution	Reagent	Flame color	Visible Atomic Spectrum Wavelength(s), nm
NaCl	K ₂ CrO ₄		
KCl			
MgCl ₂			
CaCl ₂			
SrCl ₂			
CuCl ₂			
BaCl ₂			
Unknown	—		

Data Sheet

1. Based on the Flame color what could the unknown solution be?

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